Module 4 Review

- o Know the difference between a physical change and a chemical change.
- o Know several examples of physical and chemical changes
- o Be able to classify a substance as a pure element, pure compound, homogeneous mixture, or heterogeneous mixture
- o Understand the kinetic theory of matter (how the speed of particles affects their phase)
- o Know the names of all the phase changes
- Know the list of homonuclear diatomics
- o Be able to balance a chemical equation

KNO₃

Honors:

6.

o Be able to interpret a Pressure-Temperature phase diagram

1. Fe + H₂SO₄
$$\longrightarrow$$
 Fe₂(SO₄)₃ + H₂
2. C₂H₆ + O₂ \longrightarrow H₂O + CO₂
3. KOH + H₃PO₄ \longrightarrow K₃PO₄ + H₂O
4. SnO₂ + H₂ \longrightarrow Sn + H₂O
5. NH₃ + O₂ \longrightarrow NO + H₂O

H₂CO₃

 K_2CO_3

 HNO_3